

INDIAN SCHOOL MUSCAT
CLASS :XII
SUBJECT:Physics
WORKSHEET



DATE :..... TOPIC/SUB-TOPIC :.....Atoms and Nuclei Section A Conceptual and application type questions 1 How do the neutron to Proton Ratio change during 1 i) β + ii) β - iii) α emission? 2 1 Why it is difficult to detect to neutrinos and anti neutrinos ? Define reproduction factor or multiplication factor or k factor ? 3 1 State displacement law for alpha decay . 4 1 5 What is the role of control rods in a nuclear reactor? Name the materials used in control rods 2 6 State the limitations of Rutherford 's model. 2 7 State Bohr's quantization principle and frequency condition. Define impact parameter and angle of scattering and give the relation between them. 8 2 9 What is enriched Uranium ? 2 10 3 What is the nuclear radius of ${}_{26}\text{Fe}^{125}$ if that of ${}_{13}\text{Al}^{27}$ is 3.6 F? Derive expression for the energy of electron in the nth orbits. 11 3 12 Define the distance of closest approach of an alpha particle to a gold nucleus . 3 13 What is the role of a moderator in a nuclear reactor? Explain the use of moderator in a 3 nuclear reactor?

ISM/CLASSXII/ WORKSHEET NO...11.../PHYSICS/2019-20

Section B Numerical problems

1	Calculate the binding energy and binding energy per nucleon of 26Fe56 nucleus. Given, mass of $_{26}$ Fe ⁵⁶ nucleus = 55.9349 amu mass of 1 proton = 1.007825 amu mass of 1 neutron = 1.008665 amu.	2
2	Calculate the time required for 60% of a sample of radon to undergo decay. Given $T\frac{1}{2}$ of radon = 3.8 days	2
3	The isotope ${}_{92}U^{238}$ successively undergoes three α -decays and two β -decays. What is the resulting isotope?	2
4	If 50% of a radioactive sample decays in 5 days, how much of the original sample will be left over after 20 days?	2
5	The radioactive isotope $84Po^{214}$ undergoes a successive disintegration of two α -decays and two β -decays. Find the atomic number and mass number of the resulting isotope.	2
6	Calculate the time required for 60% of a sample of radon to undergo decay. Given T ¹ / ₂ of radon = 3.8 days	2
7	The Rydberg constant for hydrogen is $1.097 \times 10^7 \text{ms}^{-1}$. Calculate the short and long wavelength limits of Lyman series.	2
8	An α – particle is projected with an energy of 4 MeV directly towards a gold nucleus. Calculate the distance of its closest approach. Given : atomic number of gold = 79 and atomic number of α particle = 2.	2
9	Wavelength of Balmer first line is 6563Å. Calculate the wavelength of second line.	2